

3-METHYL-2-PHOSPHANAPHTHALENE

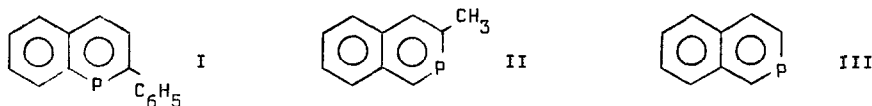
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Phosphorus analogues of benzene^{1,2}, anthracene³ and phenanthrene⁴ have been known for several years. The corresponding analogues of naphthalene, however, appeared to be less readily accessible. Recently, Märkl and Heier reported the synthesis of a derivative of 1-phosphanaphthalene, 2-phenylbenzo[b]phosphorin (I)⁵. By a different approach, namely using our previously developed method of HCl-elimination from a dihydroprecursor^{3,4} in the final step, we have prepared the first derivative of 2-phosphanaphthalene, 3-methylbenzo[c]phosphorin (II); similarly, evidence was obtained for the formation of the unsubstituted parent heterocycle III in solution.



Starting from diethyl benzylphosphonite (IVa) we obtained the ketophosphinic acid Va (m.p. 204-205° C) in analogy to Henning's synthesis of Vb from IVb⁶. Reduction of Va with NaBH₄ in H₂O and dehydration by heating with 10% H₂SO₄ yielded VI (53% yield; m.p. 200-202° C); ¹H NMR spectrum (D₆-DMSO): δ 7.35-7.10 (m, 4.5 H, arom. + 0.5 vinylic H), 7.10 (s, 1H, -OH), 6.63 (q, 0.5 H, 0.5 vinylic H), 3.12 (d, 2H, -CH₂-), 1.95 (d of d, 3H, CH₃). Reaction with thionyl chloride converted VI into its acid chloride which was reduced at -15° C by LiAlH₄ in ether to the secondary phosphine VII which was purified by molecular distillation (21% yield; b.p. ca. 80° C at 10⁻¹ Torr.); NMR spectrum (CDCl₃, external TMS): δ 7.70-7.40 (m, 4H, arom.), 7.22 (d of q, ³J_{PH} 9 Hz, ³J_{HH} 1.5 Hz, 1H, vinylic H), 4.26 (broad s, 1H, P-H), 3.32 (d, ²J_{PH} 7.5 Hz, 2H, -CH₂-), 2.54 (d of d, ³J_{PH} 12 Hz, ³J_{HH} 1.5 Hz, 3H, CH₃). The singlet for the phosphine proton is remarkable; however, its presence followed unambiguously from the IR spectrum (ca. 8% in CHCl₃), which has a P-H stretch vibration at 2250 cm⁻¹. By reaction with

phosgene in toluene (during one hour slowly warmed from -196° C to room temperature), VII was converted to the corresponding chlorophosphine which was not isolated⁷ but treated with 1,5-diazabicyclo[5.4.0]undec-5-ene (DBU) in a high vacuum sealed vessel. Filtration from DBU.HCl, evaporation of the filtrate, extraction of the residue with *n*-hexane and vacuum sublimation of the residue yielded white crystals of II (ca. 10% yield from VII, m.p. $64.5-69^{\circ}$ C).

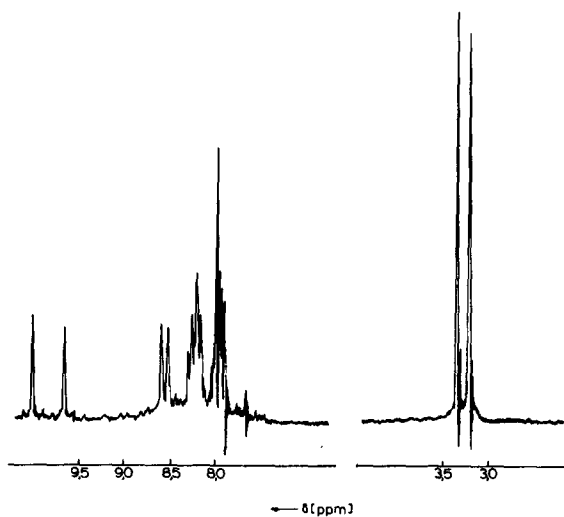


Fig. 1. 100 MHz NMR spectrum of II in CDCl_3

The structure of II follows from the elemental analysis and from its spectral data. NMR spectrum (see Fig. 1) (100 MHz, CDCl_3 , external TMS): 9.84 (d, $^2J_{\text{PH}}$ 35 Hz, 1H, H^1), 8.57 (d, $^3J_{\text{PH}}$ 7.5 Hz, 1H, H^4), 8.35-7.85 (m, 4H, benzo- C_6H_4), 3.30 (d, $^3J_{\text{PH}}$ 13.5 Hz, 3H, CH_3); chemical shifts and coupling constants are in accordance with those of Märkl⁸ and Ashe² for phosphabenzene. UV spectrum (diethyl ether): λ_{max} 252 nm ($\epsilon = 28850$), 304 nm (6630), 343 nm (450), and 360 nm (270); the spectrum disappeared rapidly on admission of air. The general agreement with UV spectra of naphthalene and isoquinoline derivatives is obvious; however, there is no simple trend in λ_{max} or in ϵ . Mass spectrum: *m/e* 160, 100% (M^+); 128, 94% [$(\text{M}-\text{PH})^+ = \text{C}_{10}\text{H}_8^+$]; 80, 12% (M^{2+}).

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